Astro 596/496 NPA Lecture 3 Aug. 28, 2009

Announcements:

• PF 1 posted due Fri. Sept. 2, \leq 12noon

Last time: abundances

- *Q: top 3 cosmic components?*
- *Q: top solar system elements?*

www: solar system abundances

clear patterns, features in solar system composition want to understand how these patterns came about

- [⊢] we will see: these patterns represent
 - a symphony of cosmic and stellar processes

Nuclear Physics

Nuclear Physics

atomic nuclei make of **nucleons** = protons and neutrons

nucleon	mass	spin	charge
proton	$m_p c^2 = 938$ MeV	$S_p = 1/2$	$Q_p = \ Q_e\ = e$
neutron	$m_n c^2 = m_p c^2 + 1.3 \text{ MeV}$	$S_n = 1/2$	$Q_n = 0$

consider a nucleus with Z protons, N neutrons mass number A = Z + N

Q: baryon number?

notation: $\begin{bmatrix} A \\ N \\ X^Z \end{bmatrix}$ but usually use shorthand, because: $_{\omega}$ chemical symbol X $\Rightarrow Z$, then can get N = A - Ze.g., $^3_1 \text{He}^2 \rightarrow {}^3 \text{He}$

www: chart of the nuclides - note axes

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isobar: fixed A (e.g., <sup>7</sup>Li and <sup>7</sup>Be)
isotope: fixed Z (e.g., <sup>6</sup>Li and <sup>7</sup>Li)
isotone: fixed N (e.g., <sup>14</sup>N and <sup>15</sup>O)
Q: any Simpson's fans?
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Q: isobar/tope/tone patterns on nuclide chart?

Nuclear Masses

to zeroth order:

mass of nucleus = $A \times$ "nucleon mass"

to make this idea precise, define **atomic mass unit (amu)**:

$$m_{\rm U} = \frac{m(^{12}{\rm C})}{12} = \frac{1 {\rm g}}{N_{\rm A}{\rm mol}} = 1.66 \times 10^{-24} {\rm g}$$

= 931.5 MeV/ $c^2 \simeq 1 {\rm GeV}/c^2$
 $\simeq m_p \simeq m_n$

(neutral atom mass: includes $6m_ec^2$)

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now define: mass excess (or mass defect)

$$\begin{array}{lll} \Delta_{i} &=& (m_{i}-A_{i}m_{u})c^{2} & (1) \\ &\uparrow \text{ neutral atom mass} & (2) \end{array}$$
e.g., $\Delta(^{12}\text{C}) = 0 \; Q: \; why?$
 $\Delta(^{16}\text{O}) = -4.737 \; \text{MeV} \\ \Delta(^{1}\text{H}) = 7.289 \; \text{MeV} \end{array}$

www: wallet card

Q: why are mass defects \neq 0 for all but ¹²C?

Nuclear Binding Energy

binding energy: for nuclide *i*

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$$B_i = \text{parts} - \text{whole}$$
(3)
= $\left(Z_i m_{\text{H}} c^2 + N_i m_n c^2\right) - m_i c^2$ (4)

stability requires $B_i > 0$, i.e., whole < sum of parts (c.f., $m_H = m_p + m_e - 13.6 \text{ eV}/c^2$) Q: why?

www: Chart of nuclides - note valley of stability

note that larger nuclei have large B_i , but shared among more nucleons, so invent: binding energy per nucleon B_i/A_i

www: Great curve of nuclear binding energy *Q: what strikes you?*

Binding Energy: Trends and Consequences

Nuclear binding energy features:

- sharp rise in B_i/A_i at low A
- local max at ⁴He
- no stable nuclei at A = 5,8
- lowest B/A for D, LiBeB
- max B/A for middle masses: peak at ⁵⁶Fe

with this in mind, revisit solar abundances: www: SS abundances

Q: significance?

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The Nuclear Fingerprint in Solar Abundances

Many observed features in solar abundances reflect observed features in nuclear binding energy curve

for example:

 ★ D, Li, Be, B are "fragile": weakly bound-low B/A but these also have very low abundances
 ★ binding has broad peak around ⁵⁶Fe

...where abundances show a broad peak

See nuclear BE effects in SS

(and stellar) abundances:

 \Rightarrow confirms: abundance pattern controlled by *nuclear physics*!

 $^{\circ}$ but atoms have their own binding too...

Q: why can't abundance pattern be due to atomic binding?

Nuclei vs Atoms

Useful to compare/contrast nuclear vs atomic interactions and structures

Q: controlling forces/interaction(s) in each?

Q: nuke/atomic similarities?

Q: nuke/atomic differences?

Properties of the Nuclear Force

EM force completely understood

via Maxwell's equations and QED

for static point charges: simple potential V(r) = Q/r

- central force
- always either attractive or repulsive

Nuclear force not fully understood, much more complicated (see, e,g., Krane, *Nuclear Physics*)

- nuclear interaction and nucleons not fundamental but manifestations of strong interactions among quarks analogy: EM force vs molecular interactions
- nucleon-nucleon (N N) potential
- not known from first principles: empirical

 $\begin{array}{c} 1\\ 1\end{array}$

The Nuclear Force

- •N N interaction *attractive* a "large" distances $\gtrsim 1$ fm, strongly *repulsive* at short distances $\lesssim 1$ fm $\rightarrow 0$ at ~ 1 fm
- N N interaction strongly spin-dependent
 e.g., d = np exists only as J = 1 (p ↑ n ↑), not J = 0 (p ↑ n ↓)
 ⇒ have s₁ · s₂ terms in N N potential
- •N N potential has *non-central* "tensor" term anisotropic, angle average = 0
- •N N force *charge symmetric*:

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n-n interaction = p-p aside from Coulomb effects

•N - N nearly *charge independent*: $V_{nn} \simeq V_{pp} \simeq V_{np}$