

Astro 210  
Lecture 11  
Feb 11, 2011

Announcements

- HW3 due
- good news: no HW next week
- bad news: Hour Exam 1 in class next Friday  
info online
- Planetarium shows: Mon, Tue, Thu next week, Mon Feb 21  
registration, report forms, info online
- Night Observing next week  
report forms, info online

## Changing Gears

Ready to begin scientific exploration of the cosmos

www: Big Picture

Need to assemble & understand available tools

What can we **directly** measure, from Earth,  
about planets/stars/galaxies?

*With partner: write list*

## Astrophysicist's Wishlist

Note that much of what we would *like* to know about celestial objects, such as

- properties: distance, size, mass, temperature, speed, spin rate, composition, ...
- physics: orbits, origin, evolution, ...

are **not** directly observable

i.e., these data aren't output of a telescope

what is?

## Observer's Toolbox

hard-nosed list of direct observables

which **do** come out of a scope:

- position on sky
- color/spectrum
- brightness
- polarization
- time changes in any/all of these above

lesson: can only measure light! can look but not touch!

⇒ need to understand light

and its interactions with matter

to decode maximum available cosmic information

amazingly lucky circumstance: can get there from here!

You can't always get what you want

No you can't always get what you want

You can't always get what you want

But if you try sometimes

You might find

You get what you need

↳ -- Astrophysicist Mick Jagger

# Electromagnetic Radiation: Wave-Particle Duality

**Heads-up:** in physics/astrophysics “radiation”  $\equiv$  EM radiation  
i.e., transport of EM energy across space by particles or waves  
 $\neq$  radioactivity = “ionizing radiation”

*Q: examples of radiation in ASTR210 sense?*

## Wave Properties

Maxwell's eqs: electric & magnetic fields\* can support waves

→ light is **electromagnetic radiation**

simplest wave: sinusoidal; more complex patterns

can be decomposed into sums of sinusoids (Fourier)

*Q: basic anatomy of any propagating sinusoidal wave?*

*Q: corresponding properties of light waves?*

*i.e., how interpreted by your personal photodetectors?*

## Electromagnetic Waves

- EM wave speed:  $c = 3.0 \times 10^8$  m/s
- spatial oscillation period: wavelength  $\lambda$
- time oscillation period:  $P$  [sec/cycle]  
related to frequency:  $f = \nu = 1/P$  [cycles/sec]
- wave travels: in time  $\Delta t = P = 1/f$ , pattern moves distance  $\Delta x = \lambda$ , and since speed is  $c = \Delta x / \Delta t \rightarrow c = \lambda f$

note: EM radiation can have any wavelength from subatomic through to macroscopic!

	radio	infrared	visible	ultraviolet	X-ray	$\gamma$ -ray
$\nu$ [Hz]	$< 10^{11}$	$\sim 10^{13}$	$\sim 5 \times 10^{14}$	$\sim 10^{16}$	$\sim 10^{18}$	$\sim 10^{20}$
$\lambda$ [m]	$> 10^{-3}$	$\sim 10^{-5}$	$\sim 5 \times 10^{-7}$ m	$\sim 10^{-9}$	$\sim 10^{-11}$	$\sim 10^{-12}$

Example: what is freq. of green light at 500 nm?

$$f = \frac{c}{\lambda} = \frac{3 \times 10^8 \text{ m/s}}{5 \times 10^{-7} \text{ m}} = 6 \times 10^{14} \text{ Hz} \quad (1)$$

## Radiation Particle Properties: Photons

leap forward: 20th century revolution of quantum mechanics

Max Planck (1858–1947):

light comes in “chunks” or “packets” of energy

→ **quantized** ⇒ **photon** (symbol  $\gamma$ )

A photon's energy set by color:  $E_\gamma = hf = hc/\lambda$

where Planck's constant  $h = 6.63 \times 10^{-34}$  Js

often also use  $\hbar = h/2\pi$



Ex: what is energy of 1 photon of green light?

$$E_{\gamma} = \frac{hc}{\lambda} \quad (2)$$

$$= \frac{6.6 \times 10^{-34} \text{ Js } 3.0 \times 10^8 \text{ m/s}}{5.0 \times 10^{-7} \text{ m}} = 4.0 \times 10^{-19} \text{ J} \quad (3)$$

$$= 2.5 \text{ eV} \quad (4)$$

very small!

new energy unit: **electron Volt**

**1 eV =  $1.602 \times 10^{-19}$  Joule**

energy gained by 1  $e^{-}$  going thru  
potential difference of 1 Volt

## iClicker Poll: Laser Pointer

*Demo:* laser pointer – emits photons all with one *same* wavelength

if I double the power output of green laser pointer beam:  
e.g., crank from 1 mWatt → 2 mWatt

What changes, what stays the same?

- A** more photons emitted, but each photon has same energy
- B** same number of photons emitted, but photon each has more energy
- C** more photons emitted *and* each photon has more energy

- each photon's  $E$  depends on  $f$  only
- **total** energy in light beam depends on  $\#$  photons

crank emitted power  $\rightarrow$  add more photons

so total energy output (power)  $\propto \#$  photons emitted per sec

# Spectroscopy

key property of light:

**flux:** energy flow per unit area per unit time

*diagram: light flow, area  $A$ ,  $dE = FAdt$ ,  $F = 1/A dE/dt$*

intuitively: “apparent brightness”

**spectrum:** flux distribution vs  $\lambda$

*Demo: use gratings*

*Q: what does white light spectrum look like?*

*Q: what does laser pointer spectrum look like?*

*Q: what's the spectrum of a neon light?*

*Q: what's the spectrum of a heated solid filament?*

can classify three basic kinds of spectra: **Kirchoff's Rules**

*diagram: hot solid, cooler gas, lines of sight*

1. A **hot and opaque** solid, liquid or dense gas emits a continuous spectrum (A)

*diagram: continuous spectrum:  $F$  vs  $\lambda$*

2. A **hot low-density (transparent) gas** produces emission line spectrum

note: pattern of lines specific to element

*diagram: emission line spectrum:  $F$  vs  $\lambda$*

3. Continuous radiation viewed through cooler gas produces an absorption line spectrum

*label C on diagram*

*diagram: absorptions line spectrum*

note: the lines absorbed have same color/wavelength as the lines in emission line spectrum:  $F$  vs  $\lambda$

these effects are godsend for astrophysics!

*Q: why?*

## Observer's Scorecard

You can see an awful lot, just by looking.

-- Astrophysicist Yogi Berra

can use emission/absorption lines to inventory  
kinds of elements in an astronomical source

light spectrum gives atom "fingerprint" or "barcode"

spectrum → composition

## Example: The Sun

Sun, stars hotter, denser in center cooler, less dense at surface  
so: sunlight/starlight shows *Q: what kind of spectrum?*

www: Sun spectrum

amount absorbed in each line → amount of atoms  
→ **composition** of Sun; works for other stars too!

Note: as yet, don't know where lines comes from  
who assigns cosmic barcodes?

for this, need to understand how light interacts with **matter**