Astro 210 Lecture 22 March 11, 2011

Announcements

- HW7 due
- Night Observing: last chance this week! first clear night today, Tue, or Wed will be *last* session due to time change, hours now 8–10 pm report forms, info online
- Hour Exam 1: scores & stats posted last night exams handed out today

Last time: finished solar system tour

Н

Today: build theory of solar system origin

#### The Age of the Earth & Solar System

Very useful, important to know age of Earth, other solar system bodies

"gold-plated" method: radioactive dating

## Radioactivity

recall: nucleus = collection of protons and neutrons not all atomic nuclei are stable! some spontaneously decay!

why? rough rule of thumb:

ω

nuclei "prefer" #n and #p roughly equal

if too many extra n or  $p \rightarrow$  change to make more even

example: Carbon-14  ${}^{14}C = \boxed{6p \ 8n}$ : 2 extra  $n \rightarrow$  unstable, radioactive decay to Nitrogen-14:  ${}^{14}N = \boxed{7p \ 7n}$ ; has equal n and p, stable how? change one  $n \rightarrow p$  *Q: why can't this be all that occurs in decay? Q: how to predict when one*  ${}^{14}C$  *nucleus decays?*  radioactive decays: to try to balance n and pdecays can change  $n \rightarrow p$ , or  $p \rightarrow n$ but note: electric charge $(p) \neq$  charge(n) $\rightarrow$  need another charged particle-electron! turns out: yet another particle too (neutrino)

for the case  $n \rightarrow p$ , really have

$$n \to p + e + \nu_e \tag{1}$$

where  $\nu_e$  is a (electron-type) neutrino (more on these later)

note:

- at deeper level, decay is quark change  $d \rightarrow u + e + \nu_e$
- $\bullet$  decay produces have high kinetic energy  $\rightarrow$  heat

## **Radioactive Decay Law**

When will a given nucleus decay? Trick question! In subatomic quantum world, decays are *random*! cannot predict when individual particle will decay! **But** can predict very accurately how a large sample will decay www: decay simulations

The rule: starting with  $n_0$  radioactive nuclei at time t = 0decay rate  $\propto$  number n of nuclei still alive  $dn/dt = -\lambda n$ 

w/ "decay constant"  $\lambda$ , depends on isotope

$$\frac{dn}{n} = -\lambda dt \tag{2}$$

$$n\left(\frac{n}{n_0}\right) = -\lambda t \tag{3}$$

СЛ

$$n(t) = n_0 e^{-\lambda t} \tag{4}$$

exponential decay law

fixed time for half of present sample to decay: half-life

rewrite:

 $n = n_0 \ 2^{-t/t_{1/2}} = n_0 \ (e^{\ln 2})^{-t/t_{1/2}} = n_0 e^{-\lambda t}$  $\rightarrow$  half life and decay rate are inverses:  $t_{1/2} = \ln 2/\lambda$ 

#### Radioactive Dating

radioactive material can be age-dated:

- ★ decay rate predictable: "clock"
- $\star$  both undecayed "parents" and decay "daughters" observable
  - $\rightarrow$  can infer amount of decay
- $\star$  some nuclei have very long  $t_{1/2} \rightarrow$  can measure very old ages

Example: Potassium-Argon dating Demo: banana

 $^{\circ}~^{40}{\rm K}$  is rare, unstable potassium isotope decays to argon  $^{40}{\rm K} \rightarrow {}^{40}{\rm Ar}$  with  $t_{1/2}=1.3\times 10^9~{\rm yr}$ 

#### Worked Example

Experiment: in rock, measure ratio  $n(^{40}\text{Ar})/n(^{40}\text{K}) = 10.6$ 

assume  $n(^{40}Ar) = 0$  at rock formation Q: why?  $\rightarrow$  what is age t of rock?

7

1. find 
$$n_0({}^{40}\mathsf{K})$$
:  

$$\frac{n({}^{40}\mathsf{Ar})}{n({}^{40}\mathsf{K})} = \frac{n_0({}^{40}\mathsf{K}) - n({}^{40}\mathsf{K})}{n({}^{40}\mathsf{K})} = 10.6$$
(5)  
 $\rightarrow n_0({}^{40}\mathsf{K})/n({}^{40}\mathsf{K}) = 11.6$ 

2. now get age:  

$$n_0({}^{40}\text{K})/n({}^{40}\text{K}) = 2^{t/t_{1/2}}$$
  
 $\Rightarrow \log_{10}(n_0/n) = t/t_{1/2}\log_{10} 2$   
 $t = \frac{\log_{10}(n_0/n)}{\log_{10} 2} t_{1/2} = 4.6 \times 10^9 \text{ yr}$ 

(6)

#### Ages of Earth and the Solar System

#### Earth

 $^{40}$ K –  $^{40}$ Ar clock "reset" whenever rocks melted Q: why?

- gives a range of dates for earth rocks *Q*: why does this make sense?
- oldest earth rocks give

$$t( ext{oldest rocks}) pprox 4.3 imes 10^9 ext{ yr } \leq t_{ ext{earth}}$$

(7)

#### Solar System

radioactive dating show: **meteorites** oldest objects strictly speaking: give *lower bound* to solar system age practically: likely formed quickly  $\rightarrow$  give SS age

 $\infty$ 

 $t_{\rm SS} = 4.6 \times 10^9 \text{ yr} = 4,600,000,000 \text{ yr} = 4.6 \text{ billion years!}$  (8)

#### **Origin of the Solar System**

theory building! recall: geocentric/heliocentric theories...

Input SS data, laws of physics

Output: Model

6

sequence of events, predictions for evolution up to present

patterns in the solar system to be explained by a theory of solar system origin *Q: similarities? differences?* 

## Solar System Data to be Explained

orbits, spins

- planet orbits in ecliptic plane
- rough spin/orbit alignments
- but some spins misaligned

Terrestrial/Jovian differences:

composition

location

size

spacing

 $_{\rm tot}$  debris: comets, asteroids

## Theory of Solar System Origin: Protosolar Nebula

stars born in cold gas & dust clumps: molecular clouds
Q: what's dust, in astro context?
www: HST Eagle Nebula

Initial protosolar material a small parcel of larger cloud

- cold gas & dust
- spinning: net angular momentum  $\neq 0$ Q: why is  $\vec{L} \neq 0$  a reasonable assumption?

For simplicity: imagine first a cold cloud with *zero* spin

i.e., *zero* angular momentum

Q: forces on particles in cloud?

*Q: response of particles to these forces?* 

11

Q: why is coldness important for this to work?

# **Gravitational Collapse**

ignoring spin:

particles in cold cloud feel forces of

- gravity
- thermal pressure

but if cloud is cold: T low, pressure  $P = \rho kT/m_{\text{particle}}$  small  $\rightarrow$  only important force is gravity

diagram: cloud  $\rightarrow$  collapse gravity  $\rightarrow$  inward motion  $\rightarrow$  denser  $\rightarrow$  stronger gravity  $\rightarrow$  runaway! "gravitational collapse"

12

Q: why doesn't collapse continue until all matter  $\rightarrow$  point?

# iClicker Poll: Contraction of a Spinning Swarm

Consider a swarm of particles, spinning around an axis Which is easier?

- A moving a particle *parallel* to spin ↓ toward midplane
- B moving a particle *perpendicular* to spin
  - $\leftarrow$  toward spin axis
- C both motions equally easy

## Nebular Collapse: Birth of Sun and Disk

indeed, most matter compressed  $\rightarrow$  central "proto-Sun"

but real pre-stellar clouds are clumpy parts of larger nebulae  $\rightarrow$  turbulent motions

 $\rightarrow$  clumps have random but nonzero spins:  $\vec{L}_{init} \neq 0$ 

spin  $\rightarrow$  axial but not spherical symmetry describe with cylindrical coordinates  $(r, \theta, z)$ 

 $\rightarrow$  collapse not spherical

angular momentum "centrifugal barrier" along R, but not z

 $\Rightarrow$  collapse easier along z

 $\Rightarrow$  protoplanetary disk

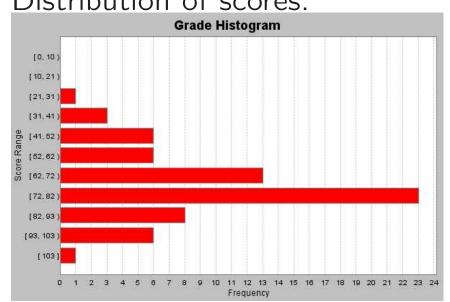
diagram: disk

14

disk  $\rightarrow$  planet & debris orbit planes, spin axes

# Hour Exam 1

- Scores and statistics posted on Compass.
- median = 74, std deviation = spread around mean = 16.8
- Solutions posted online.



#### Distribution of scores:

#### 15

Recall:

- this exam worth 10% of final grade
- equivalent to 2 HW grades