

# Astro 501: Radiative Processes

## Lecture 5

Jan 25, 2013

Announcements:

- **Problem Set 1** due **now**
- **Problem Set 2** available, due at start of class next Friday

Last time: the glorious equation of radiation transfer

*Q: what is it?*

*Q: what is optical depth? column density?*

*Q: what is source function? why is it important?*

equation of radiation transfer

$$\frac{dI_\nu}{ds} = -\alpha_\nu I_\nu + j_\nu = -\alpha_\nu (I_\nu - S_\nu)$$

$$\frac{dI_\nu}{d\tau_\nu} = -I_\nu + S_\nu$$

with source function

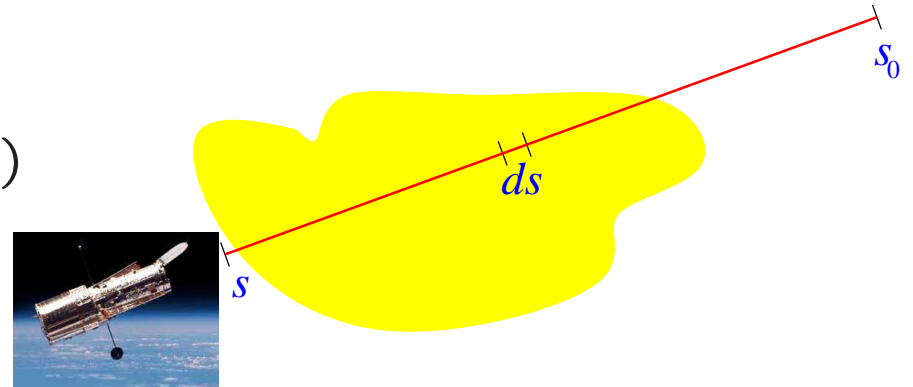
$$S_\nu = \frac{j_\nu}{\alpha_\nu n u} \quad (1)$$

and optical depth  $d\tau_\nu = \alpha_\nu ds$ , so that

$$\tau_\nu = \int_{s_0}^s \alpha_\nu ds = \sigma_\nu N_a \quad (2)$$

with column density

$$N_a = \int_{s_0}^s n_a ds \quad (3)$$



# Blackbody Radiation

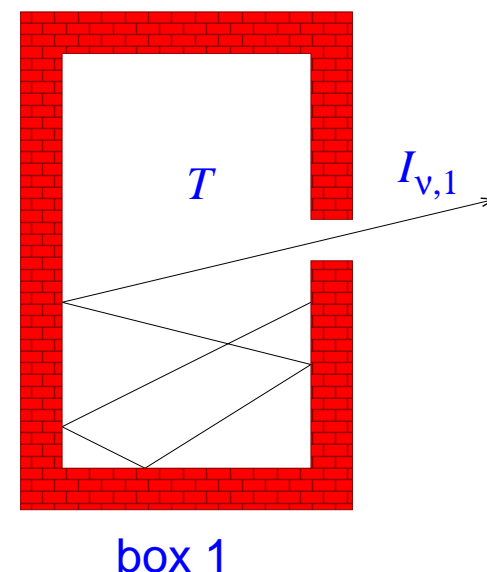
# Radiation and Thermodynamics

consider an enclosure ( “*box 1*” )  
in *thermodynamic equilibrium* at temperature  $T$

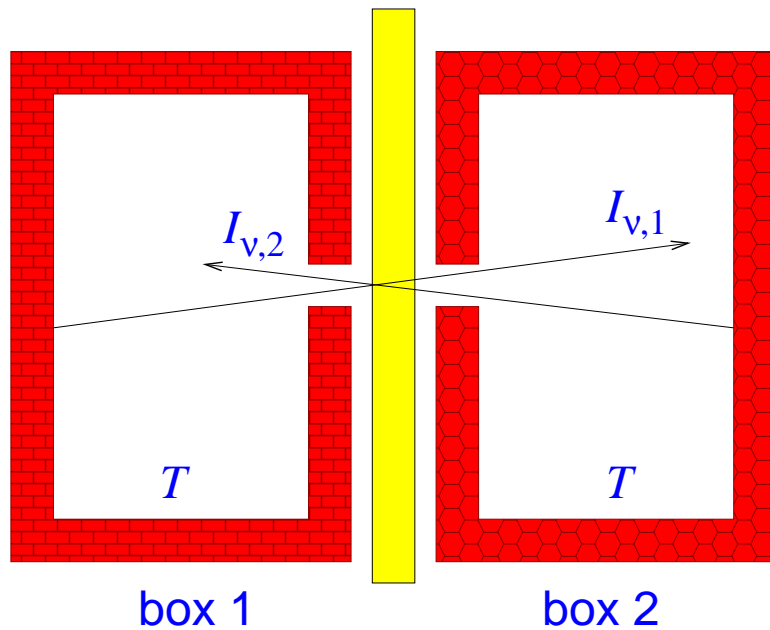
the matter in box 1

- is in random thermal motion
- will absorb and emit radiation  
details of which depends on  
the details of box material and geometry
- but equilibrium  
→ radiation field in box doesn't change

open little hole: escaping radiation has intensity  $I_{\nu,1}$



now add another enclosure ( “box 2” ), also at temperature  $T$  but made of *different material*



separate boxes by *filter passing only frequency  $\nu$*   
radiation from each box incident on screen

Q: imagine  $I_{\nu,1} > I_{\nu,2}$ ; what happens?

Q: lesson?

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Q: how would  $I_{\nu,1}$  change if we increased the box volume  
but kept it at  $T$ ?

# Blackbody Radiation

if both boxes at *same  $T \Rightarrow$  no net energy transfer*

but this requires  $I_{\nu,1} = I_{\nu,2}$  and so the radiation is:

- independent of the composition of the box
- a universal function of  $T$
- isotropic  $Q$ : *why?*
- **blackbody radiation** with intensity  $B_{\nu}(T)$

Implications:

- $B_{\nu}(T)$  and thus  $B(T)$  depends only on  $T$ , not on cavity volume  $V$  or composition
- thus blackbody energy density  $u(T) = 4\pi B(T)/c$  also depends only on  $T$ , not on  $V$
- thus in volume  $V$ , photon energy is  $U = u(T) V$
- and pressure is  $P(T) = u(T)/3$ , also independent of  $V$

○

Lesson: radiation has energy, exchanges it with environment

→ *radiation can be treated thermodynamically*

# Thermodynamics Recap

**First Law of Thermodynamics:** heat is work!  
adding *heat energy*  $dQ$  to system changes  
system *energy*  $U$  and/or *pressure*  $P$ :

$$dQ = dU + p dV \quad (4)$$

**Second Law of Thermodynamics:** heat is entropy!

$$T dS = dQ \quad (5)$$

together

$$T dS = dU + P dV \quad (6)$$

and thus entropy  $S = S(T, V)$  obeys

$$dS = \frac{dU}{T} + \frac{P}{T} dV \quad (7)$$

entropy  $S = S(T, V)$  obeys

$$dS = \frac{dU}{T} + \frac{P}{T}dV \quad (8)$$

and thus we have

$$\partial_T S = \frac{\partial_T U}{T} \quad (9)$$

$$\partial_V S = \frac{\partial_V U + P}{T} \quad (10)$$

which means

$$\partial_V \partial_T S = \frac{\partial_V \partial_T U}{T} \quad (11)$$

$$\partial_T \partial_V S = \frac{\partial_T \partial_V U}{T} - \frac{\partial_V U}{T^2} + \partial_T \left( \frac{P}{T} \right) \quad (12)$$

but mix partial derivatives equal, e.g.,  $\partial_V \partial_T S = \partial_T \partial_V S$ ,  
and note that  $\partial_V U|_T = u$  energy density, so

$\infty$

$$u = T^2 \partial_T \left( \frac{P}{T} \right) \quad (13)$$



## Radiation Thermodynamics

general thermodynamic considerations give:

$$u = T^2 \partial_T \left( \frac{P}{T} \right) \quad (14)$$

now specialize to *radiation*:  $P = P(T) = u(T)/3$

$$T \frac{d}{dT} \left( \frac{u}{T} \right) = 3 \frac{u}{T} \quad (15)$$

which gives

$$\frac{d(u/T)}{u/T} = 3 \frac{dT}{T} \quad (16)$$

$$\ln \left( \frac{u}{T} \right) = 3 \ln(T) + \ln(a) \quad (17)$$

$$u(T) = a T^4 \quad (18)$$

radiation energy density

$$u(T) = a T^4 \quad (19)$$

- $u(T) \propto T^4$ : strong  $T$  dependence!
- implies  $B(T) = ac/4\pi T^4$ ,  
and  $F(T) = \pi B(T) = ac/4 T^4$
- $a$  is the “radiation constant”  
value not determined by thermodynamics alone

Note: *blackbody quantities fixed entirely by  $T$*   
no adjustable parameters!

## Radiation Entropy

Using  $U = aT^4V$  and  $P = u/3$ , can solve for **radiation entropy**

$$S_{\text{rad}} = \frac{4}{3}aT^3 V \quad (20)$$

and thus **entropy density**  $s_{\text{rad}}(T) = S/V = 4/3 aT^3$

if entropy constant in a parcel of radiation  
→ **adiabatic** process:

$$T_{\text{adiabat}} \propto V^{-1/3} \quad (21)$$

$$P_{\text{adiabat}} \propto T_{\text{adiabat}}^4 \propto V^{-4/3} \quad (22)$$

writing  $P \propto V^{-\gamma}$ , we have  
an **adiabatic index**  $\gamma_{\text{rad}} = 4/3$

*Q: but how do we get the constant  $a$ ?*

# Gossip Break: Chandra Story

# The Quantum Mechanics of Blackbody Radiation

to have deeper understanding of radiation thermodynamics  
and to find radiation constant  $a$   
need to study radiation in more detail  
→ need physical picture of radiation

can try classical description: radiation as EM waves  
different frequencies ( “modes” ) all thermally excited  
→ gives somewhat wrong answers, e.g.,  $u(T) = 8\pi kT/c^3 \int_0^\infty \nu^2 d\nu \rightarrow \infty$   
“ultraviolet catastrophe”

Historically, this disaster drove Planck & Einstein to a new  
*microscopic* picture of quanta: photons

→ of course this gives correct blackbody description  
in a *statistical mechanics* description of photons

# Statistical Mechanics in a Nutshell

classically, **phase space**  $(\vec{x}, \vec{p})$   
completely describes particle state

*Q: phase space lifestyle of single classical 1-D free body?  
of single 1-D harmonic oscillator?*

*Q: a swarm of free bodies? oscillators?*

but quantum mechanics  $\rightarrow$  uncertainty  $\Delta x \Delta p \geq \hbar/2$

semi-classically:

can show that a quantum particle must occupy  
a **minimum** phase space “volume”

<sup>14</sup>  $(dx \ dp_x)(dy \ dp_y)(dz \ dp_z) = h^3 = (2\pi\hbar)^3$   
per quantum state of fixed  $\vec{p}$

## Distribution Function

define “occupation number” or “distribution function”  $f(\vec{x}, \vec{p})$ :  
number of particles in each phase space “cell”

*Q:  $f$  range for fermions? bosons?*

*Q: what is  $f$  for one classical particle? many classical particles?*

Given distribution function, total number of particles is

$$dN = g f(\vec{x}, \vec{p}) \frac{d^3\vec{x} d^3\vec{p}}{h^3} \quad (23)$$

where  $g$  is # internal (spin/helicity) states:

*Q:  $g(e^-)$ ?  $g(\gamma)$ ?  $g(p)$ ?*

particle phase space occupation  $f$  determines bulk properties

*Q: how? Hint—what’s # particles per unit spatial volume?*

Fermions:  $0 \leq f \leq 1$  (Pauli)

Bosons:  $f \geq 0$   $g(e^-) = 2s(e^-) + 1 = 2$  electron, same for  $p$   
 $g(\gamma) = 2$  (*polarizations*) *photon*

Particle phase space occupation  $f$  determines bulk properties

Number density

$$n(\vec{x}) = \frac{d^3N}{d^3x} = \frac{g}{h^3} \int d^3\vec{p} f(\vec{p}, \vec{x}) \quad (24)$$

*Q: this expressions is general—specialize to photons?*



for photons  $E = cp = h\nu$

so  $d^3p = p^2 dp d\Omega = h^3/c^3 \nu^2 d\nu d\Omega$

photon number density is thus

$$dn = \frac{2}{c^3} \nu^2 f(\nu) d\nu d\Omega \quad (25)$$

and thus we have

$$\frac{dn_\nu}{d\Omega} = \frac{dn}{d\nu d\Omega} = \frac{2}{c^3} \nu^2 f(\nu) \quad (26)$$

thus *f gives a general, fundamental description of photon fields*

the challenge is to find the physics that determines  $f$

→ spoiler alert: you have already seen a version of it!

but will see it again as the Boltzmann equation!

Note: distribution function  $f(\nu)$  and specific intensity  $I_\nu$  are *equivalent* and *interchangeable descriptions*

Q: why? how do we get  $I_\nu$  from  $f(\nu)$ ?

## Distribution Function and Observables

distribution function  $f(\nu)$  is related to photon number via

$$\frac{dn_\nu}{d\Omega} = \frac{dN}{dV d\nu d\Omega} = \frac{2}{c^3} \nu^2 f(\nu) \quad (27)$$

but we found that photon specific intensity is related to specific number density via

$$I_\nu = c h\nu \frac{dn_\nu}{d\Omega} \quad (28)$$

but this means that the two are related via

$$I_\nu = \frac{2h}{c^2} \nu^3 f(\nu) \quad (29)$$

## Equilibrium Occupation Numbers

So far, totally general description of photon fields  
no assumption of thermodynamic equilibrium

in thermodynamical equilibrium at  $T$ , the distribution function  
is also the *occupation number*

i.e., average *number* of photons with freq  $\nu$

$$f(\nu, T) = \frac{1}{e^{h\nu/kT} - 1} \quad (30)$$

see derivation in today's Director's Cut Extras

Q: at fixed  $T$ , for which  $\nu$  is  $f$  large? small?

Q: sketch of  $f(\nu)$ ?

Q: what does this all mean physically?

Q: when is  $f$  zero?

Q: in which regime do we expect classical behavior? quantum?

## Blackbody Radiation Properties

Using the blackbody distribution function, we define

$$B_\nu(T) \equiv I_\nu(T) = \frac{2h}{c^2} \nu^3 f(\nu, T) \quad (31)$$

and thus we have

$$B_\nu(T) = \frac{2h}{c^2} \frac{\nu^3}{e^{h\nu/kT} - 1} \quad (32)$$

with  $h$  = Planck's constant,  $k$  = Boltzmann's constant

in wavelength space

$$B_\lambda(T) = 2hc^2 \frac{\lambda^{-5}}{e^{hc/\lambda kT} - 1} \quad (33)$$

# Director's Cut Extras

## Blackbody Photon Occupation Number

at a fixed temperature  $T$  and frequency  $\nu$   
we want the distribution function  $f$ , i.e., the occupation number  
i.e., the **average number** of photons with frequency  $\nu$

Boltzmann: probability of having state  $n$  of energy  $E_n$   
proportional to  $p_n = e^{-E_n/kT}$

Planck:  $n$  photons have  $E_n = h\nu$ , so  $p_n = e^{-nx}$   
with  $x = h\nu/kT$

So average number is

$$f = \langle n \rangle = \frac{\sum_n n p_n}{\sum_n p_n} = \frac{\sum_n n e^{-nx}}{\sum_n e^{-nx}} \quad (34)$$

note that  $\sum_n n e^{-nx} = -\partial_x \sum_n e^{-nx}$ , so

$$f = -\partial_x \ln \left( \sum_n e^{-nx} \right) \quad (35)$$

but geometric series has sum

$$\sum_n e^{-nx} = \sum_n (e^{-x})^n = \frac{1}{1 - e^{-x}} \quad (36)$$

and thus

$$f = -\partial_x \ln \frac{1}{1 - e^{-x}} = \partial_x \ln(1 - e^{-x}) \quad (37)$$

$$= \frac{e^{-x}}{1 - e^{-x}} \quad (38)$$

which gives

$$f(\nu, T) = \frac{1}{e^{h\nu/kT} - 1} \quad (39)$$

which was to be shewn